Synthesis, Characterization and Biological evaluation of some novel Pyrazolo [1,5-a]Pyrimidine derivatives

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ABSTRACT: A convenient synthesis of substituted Pyrazolo[1,5-a]pyrimidine was carried out by the reaction of different ketene dithioacetals with different aromatic amine in isopropanol in the presence of potassium carbonate. The newly synthesized compound were characterized by ¹H NMR, IR, Mass and screened for their antimicrobial activity against various strains of bacteria and fungi. From the synthesized different NCEs, compounds 8a, 8d and 8e are broad spectrum drug which can inhibit the growth of gram positive, gram negative bacteria and fungi.

1. INTRODUCTION

Functionalized nitrogen heterocycles play a prominent role in medicinal chemistry and therefore they have been intensively used as scaffolds for drug development [1]. Various heterocycles such as pyrazole, pyridine and pyrimidine have been used as key pharmacophores. Pyrazolo[1,5-a] pyrimidines have attracted considerable interest because of their biological activity. The heterocyclic fusion of pyrimidine ring and pyrazole ring resulted in formation of pyrazolopyrimidines, the structural analogues of biogenic purine class, undoubtedly, has high significance in the field of pharmaceutical and biotechnological sciences with wide spectrum of biological activities and its several derivatives [2-4]. Several compounds of this class display interesting antitrypanosomal [4] and antischistosomal activities [6]. They are used as HMG-CoA reductaseinhibitors [7], COX-2 selective inhibitors [8], 30,50-cyclic-AMP phosphodiesteraseinhibitors [9], CRF1 antagonists [10-11], selective peripheral benzodiazepine receptor ligands [12-16], potassium channel [17] and histamine-3 receptor ligands [18], CDK9 inhibitor [19]and antianxiety agents [20].

Some pyrazolopyrimidines serve as efficient sedative-hypnotic and anxiolytic drugs like zaleplon (Sonata, hypnotic) [21], indiplon (hypnotic) [22-24] and ocinaplon (anxiolytic)[25-26], fasiplon (anxiolytic). These drugs are related to the class of nonbenzodiazepines, and their therapeutic effect is due to allosteric enhancement of the action of the inhibitory neurotransmitter GABA at the GABAA receptor. These examples emphasize the importance of pyrazole-fused heterobiaryls, as well as pyrazolopyridines, as key pharmacophores in bioactive small molecules.

We have synthesize pyrazolo[1,5-a] pyrimidine derivatives by refluxing different ketene dithioacetals with aromatic amine in the presence of potassium carbonate in isopropanol.

The newly synthesized compounds were characterized by IR, Mass and ¹H NMR. All the synthesized compounds were evaluated for their antimicrobial activity.

Scheme 1: Reaction Scheme for 8a-l

2. EXPERIMENTAL

All chemicals used were commercial supply and used without further purification. The progress of the reaction was monitored by analytical TLC on precoated plates (silica gel 60, F254) and visualized with UV light. Melting points were determined using open capillary tube and are uncorrected. NMR spectra (^1H at 400 MHz) were recorded using DMSO- d_6 as a solvent and chemical shifts are expressed in parts per million (ppm) related to internal standard TMS. Infrared spectra were determined on a Shimadzu FT-IR. The specifications of the LC/MS are as follows: electrospray (+) ionization, mass range 100-800 Da, 20-V cone voltage, and Xterra MS C_{18} column (2.1 mm x 50 mm x 3.5 μm).

General procedure for the Preparation of 2-Cyano-N-arylacetamide (2)

To a stirred solution of different aromatic amine (110 mmol) **1** and cyanoacetic acid (121 mmol) in the presence of triethylamine (165 mmol) in dichloromethane (100 ml), EDC.HCl (121 mmol) was added lotwise at 0-10°C with 30 to 60 minutes. The reaction mixture was allowed to stir at room temperature for 2 to 3 hours. The progress of the reaction was monitored on TLC. After completion of the reaction water was added into the reaction mixture. The organic layer was separated and washed with water and dried over sodium sulphate. The solvent was distilled under vacuum to give 2-Cyano-N-arylacetamide **2** as a cream to brown color solid compound which was used for the next stage without purification (82-89 % yield).

General procedure for the Preparation of 2-cyano-3,3-bis(methylthio)-N-arylacrylamide (3)

To a stirred solution of 2-cyano-N-arylacetamide (20 mmol) **2** in N,N-dimethylformamide (25 ml), dry potassium carbonate (20 mmol) was added at room temperature. The reaction mixture was stirred at room temperature for 2 hours. Carbon disulfide (75 mmol) was added at room temperature and the reaction mixture was stirred for 2 hours. Then the reaction mixture was cooled to 0-10°C and methyl iodide (50 mmol) was added in to the reaction mixture and the reaction mixture was stirred for 3 to 4 hours at room temperature. The progress of the reaction was monitored on TLC. After completion of the reaction, water was added into the reaction mixture and the reaction mixture was stirred for 1 hour at room temperature. The precipitated solid material was filtered, washed with water and dried to give 2-Cyano-3,3-bis(methylthio)-N-arylacrylamide **3** (75-88 % yield) as a yellow solid.

General procedure for the preparation of 5-amino-N-aryl-3-(methylthio)-1H-pyrazole-4-carboxamide (4)

To a stirred solution of 2-Cyano-3,3-bis(methylthio)-N-arylacrylamide (20 mmol) **3** and hydrazine hydrate (25 mmol) in isopropanol (100 ml) was heated to reflux for 2 to 3 hours. The progress of the reaction was monitored by TLC. After completion of the reaction, the reaction mixture was cooled to room temperature. The reaction mixture was poured into crushed ice and stirred for 1 hour at room temperature. The precipitated solid material was filtered, washed with water and dried to give 5-amino-N-aryl-3-(methylthio)-1H-pyrazole-4-carboxamide **4** as a yellow color solid compound in 70-90 % yield.

General procedure for the preparation of different Acetoacetanilide derivatives (6)

To a stirred solution of different aromatic amine (10 mmol) **5** and ethylacetoacetate (20 mmol) in the presence of catalytic amount potassium hydroxide (10%) in toluene (100 ml) was heated at reflux temperature for 15 to 20 hours. The progress of the reaction was monitored by TLC. After completion of the reaction the solvent was removed under vacuum and methanol was added into the residue and stirred for 1 hour at room temperature. The precipitated solid material was filtered, washed with methanol and dried to give different acetoacetanilide compounds **6** as an off white to light yellow color solid compound in 35-45 % yield.

General procedure for the preparation of different ketene dithioacetals (7)

To a stirred solution of acetoacetanilide derivatives (10 mmol) **6** in N,N-dimethylformamide (10 ml), dry potassium carbonate (11 mmol) was added at room temperature. The reaction mixture was stirred at room temperature for 2 hours. Carbon disulfide (30 mmol) was added at room temperature and the reaction mixture was stirred for 2 hours. Then the reaction mixture was cooled to 0-10°C and methyl iodide (25 mmol) was added in to the reaction mixture and the reaction mixture was stirred for 3 to 4 hours at room temperature. The progress of the reaction was monitored by TLC. After completion of the reaction, water was added into the reaction mixture and the reaction mixture was stirred for 1 hour at room temperature. The precipitated solid material was filtered, washed with water and dried to give ketene dithioacetals derivatives **7** as a yellow color solid in 70-80% yield.

General procedure for the preparation of fused pyrazolopyrimidines (8)

To a stirred mixture of 5-amino-N-aryl-3-(methylthio)-1Hpyrazole-4-carboxamide (5 mmol) **4**, ketene dithioacetals (5 mmol) **7** potassium carbonate (10 mmol) in (10 ml) was heated to reflux temperature for 12 to 14 hours. The progress of the reaction was monitored by TLC. After completion of reaction the reaction mixture was cooled to room temperature and water was added. The reaction mixture was stirred for 1 hour at room temperature. The solid material was filtered, washed with water and dried to give fused pyrazolopyrimidine derivatives. The compound was crystallized from methanol to give pure compound in 80-90% yield.

Table 1: Physical Data of Compound **8a-l**

Compound Code	R	R ₁	R ₂	M.F.	Yield (%)
8a	H	H	O	C ₂₇ H ₂₆ N ₆ O ₄ S ₂	85
8b	4-CH ₃	H	O	C ₂₈ H ₂₈ N ₆ O ₄ S ₂	88
8c	4-OCH ₃	H	O	C ₂₈ H ₂₈ N ₆ O ₅ S ₂	83
8d	4-Cl	H	O	C ₂₇ H ₂₅ ClN ₆ O ₄ S ₂	80
8e	H	H	H ₂	C ₂₇ H ₂₈ N ₆ O ₃ S ₂	82
8f	4-CH ₃	H	H ₂	C ₂₈ H ₃₀ N ₆ O ₃ S ₂	90
8g	4-OCH ₃	H	H ₂	C ₂₈ H ₃₀ N ₆ O ₄ S ₂	84
8h	4-Cl	H	H ₂	C ₂₇ H ₂₇ ClN ₆ O ₃ S ₂	85
8i	H	F	H ₂	C ₂₇ H ₂₇ FN ₆ O ₃ S ₂	87
8j	4-CH ₃	F	H ₂	C ₂₈ H ₂₉ FN ₆ O ₃ S ₂	90
8k	4-OCH ₃	F	H ₂	C ₂₈ H ₂₉ FN ₆ O ₄ S ₂	85
8l	4-Cl	F	H ₂	C ₂₇ H ₂₆ ClFN ₆ O ₃ S ₂	80

2. 1. Spectroscopic data for the Compounds:

7-methyl-2,5-bis(methylthio)-N3-(4-morpholinophenyl)-N6-(p-tolyl)pyrazolo[1,5-a]pyrimidine-3,6-dicarboxamide (**8f**)

Yield: 90 %. Yellow solid; mp >300°C; R_f 0.43 (9:1 DCM-Methanol); ¹H NMR (400 MHz, DMSO-d₆): δ= 2.271 (s, 3H), 2.788 (s, 3H), 3.045-3.068 (t, 4H), 3.731-3.754 (t, 4H), 6.925-6.948 (d, 2H), 7.114-7.136 (d, 2H), 7.531-7.571 (dd, 4H), 10.617 (s, 1H), 12.113 (s, 1H) ppm; **MS:** m/z 563.1 (M+1)⁺; **IR (KBr) Cm⁻¹:** 3030.17, 2970.38, 1734.01, 1716.65, 1539.20, 1508.33, 1228.66, 1112.93, 817.82

N6-(4-chlorophenyl)-N3-(3-fluoro-4-morpholinophenyl)-7-methyl-2,5-bis(methylthio)pyrazolo[1,5-a]pyrimidine-3,6-dicarboxamide (**8l**)

Yield: 80 %. Yellow solid; mp >300°C; R_f 0.41 (9:1 DCM-Methanol); ¹H NMR (400 MHz, DMSO-d₆): δ= 2.785 (s, 3H), 2.951-2.972 (t, 4H), 3.709-3.753 (t, 4H), 7.006-7.052 (t, 1H), 7.206-7.234 (dd, 1H), 7.358-7.380 (dd, 2H), 7.704-7.748 (m, 3H), 10.784 (s, 1H), 12.252 (s, 1H) ppm; **MS:** m/z 601.1 (M(35Cl)+1)⁺, m/z 603.0 (M(37Cl)+1)⁺; **IR (KBr) Cm⁻¹:** 3047.58, 2970.38, 1734.01, 1716.65, 1539.20, 1508.33, 1226.73, 1111.00, 813.96

N3-(3-fluoro-4-morpholinophenyl)-N6-(4-methoxyphenyl)-7-methyl-2,5-bis(methylthio)pyrazolo[1,5-a]pyrimidine-3,6-dicarboxamide (**8k**)

Yield: 85 %. cream solid; mp >300°C; R_f 0.49 (9:1 DCM-Methanol); ¹H NMR (400 MHz, DMSO-d₆): δ= 2.784 (s, 3H), 2.950-2.972 (t, 4H), 3.709-3.739 (m, 7H), 6.832-6.915 (d, 2H), 7.005-7.051 (t, 1H), 7.202-7.224 (d, 1H), 7.579-7.602 (d, 2H), 7.705-7.748 (dd, 1H), 10.812 (s, 1H), 11.967 (s, 1H) ppm; **MS:** m/z 596.2 (M+1)⁺; **IR (KBr) Cm⁻¹:** 3030.17, 2929.87, 1734.01, 1716.65, 1539.20, 1521.84, 1247.94, 1111.00, 810.10

Table 2: Antibacterial activity of compound **8a-l**

Compounds	Antibacterial MIC ($\mu\text{g/mL}$)				Antifungal MIC ($\mu\text{g/mL}$)	
	<i>B. megaterium</i> MTCC2444	<i>S. aureus</i> MTCC737	<i>E. coli</i> MTCC1687	<i>P. aeruginosa</i> MTCC3541	<i>A. niger</i> MTCC282	<i>A. flavus</i> MTCC418
Streptomycin			50	50		
Ampicillin	100	100				
Nystatin					100	100
8a	500	500	500	125	125	500
8b	1000	1000	1000	1000	1000	1000
8c	1000	1000	1000	1000	1000	1000
8d	500	500	500	500	500	500
8e	500	500	500	500	250	250
8f	500	500	500	500	1000	500
8g	500	1000	250	500	500	500
8h	500	500	500	1000	1000	500
8i	1000	500	500	1000	1000	1000
8j	1000	1000	1000	1000	1000	1000
8k	500	500	500	250	250	250
8l	1000	500	500	250	250	500

Fig. 1: Antibacterial activity chart of compound **8a-l**

3. RESULT AND DISCUSSION

Chemistry: The synthesis of 7-Methyl-2,5-bis(methylthio)-N³,N⁶-diarylpyrazolo[1,5-a]pyrimidine-3,6-dicarboxamide is outlined in scheme-1. Different aromatic amine is reacted with ethylacetoacetate in the presence of catalytic sodium hydroxide in toluene to give acetanilide derivatives, which is reacted with carbon disulfide the presence of base followed by methyl iodide in N,N-dimethyl formamide to give ketene dithioacetyl derivatives. Different aromatic amine is reacted with cyano acetic acid in the presence of EDC.HCl in dichloromethane to give cyano acetamide derivative, which on reaction with carbon disulfide in the presence of base followed by methyl iodide to yield ketene dithiocyanoacetyl derivatives. Ketene dithiocyanoacetal derivatives cyclized with hydrazine hydrate to give amino pyrazole derivative. Different amino pyrazole derivatives is reacted with different ketene dithioacetyl derivatives to yield different pyrazolo [1,5-a]pyrimidine derivatives.

Biological: All the synthesized compounds were screened against varieties of bacterial strains such *Bacillus megaterium*, *Staphylococcus aureus*, *Escherichia coli*, *Pseudomonas aeruginosa*, *Aspergillus niger* and *Aspergillus flavus* strains at minimal inhibitory concentration (MIC). Standard drugs like Streptomycin, Ampicillin and Nystatin were used for the comparison purpose. Compounds **8a**, **8d** and **8e** are broad spectrum drug which can inhibit the growth of gram positive, gram negative bacteria and fungi. Among these compounds compound **8a** is more potent. Compound **8g** showed good activity against *Escherichia coli*, moderate activity against *Bacillus megaterium* & *Pseudomonas aeruginosa* and poor activity against *Staphylococcus aureus*. Compound **8h** displayed moderate activity against *Bacillus megaterium*, *Staphylococcus aureus* and *Escherichia coli*, while poor activity against *Pseudomonas aeruginosa*. Compound **8i** showed moderate activity against *Staphylococcus aureus* & *Escherichia coli* and poor activity against *Bacillus megaterium* & *Pseudomonas aeruginosa*. Compound **8j** displayed food activity against *Pseudomonas aeruginosa* and moderate activity against other strains. Compound **8l** showed good activity against *Pseudomonas aeruginosa* and moderate activity against *Staphylococcus aureus* & *Escherichia coli* and poor activity against *Bacillus megaterium*, while other showed moderate to poor activity.

Minimal fungicidal activity showed that compound **8e** and **8l** displayed good activity against *Aspergillus niger* and *Aspergillus flavus* while compound **8d** and **8g** showed moderate activity against the same. Compound **8a** showed good activity against *Aspergillus niger* and moderate activity against *Aspergillus flavus*. Compound **8f** and **8h** displayed moderate activity against *Aspergillus flavus* Compound, while remaining all possessed poor activity against all fungal stains.

4. CONCLUSION

An efficient method for preparing substituted pyrazolo [1,5-a]pyrimidine derivatives was described and the structure of synthesized compounds was determine by IR, ¹H NMR, and Mass spectroscopic analysis and evaluated for their in vitro antimicrobial activity by broth dilution method which shows good to poor activity against different bacterial strains.

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Representative Spectra:

 ^1H NMR (DMSO- d_6) spectra of compound 8f

Fig. 2a

Fig. 2b

Fig. 2c

^1H NMR (DMSO- d_6) spectra of compound 8i

Fig. 3a

Fig. 3b

Fig. 3c

^1H NMR (DMSO- d_6) spectra of compound 8l

Fig. 4a

Fig. 4b

Fig. 4c

Fig. 4d

Appendix:

Table 3: Structure of synthesized NCEs:

Compound Code	Structure	Molecular formula
8a		$C_{27}H_{26}N_6O_4S_2$
8b		$C_{28}H_{28}N_6O_4S_2$
8c		$C_{28}H_{28}N_6O_5S_2$
8d		$C_{27}H_{25}ClN_6O_4S_2$

8e		$C_{27}H_{28}N_6O_3S_2$
8f		$C_{28}H_{30}N_6O_3S_2$
8g		$C_{28}H_{30}N_6O_4S_2$
8h		$C_{27}H_{27}ClN_6O_3S_2$

8i		$C_{27}H_{27}FN_6O_3S_2$
8j		$C_{28}H_{29}FN_6O_3S_2$
8k		$C_{28}H_{29}FN_6O_4S_2$
8l		$C_{27}H_{26}ClFN_6O_3S_2$

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